

Preliminary communication

DITHIOFORMATO COMPLEXES OF THE PLATINUM METALS; SYNTHESIS AND PROTON NMR SPECTRA

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Summary

Proton NMR spectra of the dithioformate ligands in an extensive range of new platinum metal dithioformate complexes permit unambiguous assignment of stereochemistry.

The stereochemistry, structure and even the stoichiometry of products obtained on treatment of carbon disulphide with transition metal complexes are frequently subjects of doubt and speculation [1]. We now find that the dithioformate anion, one of the ligands commonly found in these systems [1-3], can be reliably detected and characterised by its IR and proton NMR spectra. Furthermore the NMR couplings of the dithioformate proton (HCS_2) to the hydride [$^4J(\text{HH})$] and/or phosphorus donor ligands [$^4J(\text{PH})$] are valuable sources of stereochemical information. To confirm these observations we have synthesised and characterised the most extensive series of dithioformate complexes reported to date.

The hydrides $[\text{MHX}(\text{CO})(\text{PPh}_3)_3]$ ($\text{M} = \text{Ru, Os; X} = \text{Cl, Br}$) $[\text{MH}(\text{OCOCF}_3)(\text{CO})(\text{PPh}_3)_2]$, $[\text{RuH}_2(\text{PPh}_3)_4]$, $[\text{OsH}_4(\text{PPh}_3)_3]$, $[\text{IrHCl}_2(\text{PPh}_3)_3]$ (*trans* chlorides) and *mer*- $[\text{IrH}_3(\text{PPh}_3)_3]$ react with carbon disulphide in boiling benzene to yield dithioformate derivatives $[\text{M}(\text{S}_2\text{CH})\text{X}(\text{CO})(\text{PPh}_3)_2]$ (two isomeric forms), $[\text{M}(\text{S}_2\text{CH})(\text{OCOCF}_3)(\text{CO})(\text{PPh}_3)_2]$, $[\text{Ru}(\text{S}_2\text{CH})_2(\text{PPh}_3)_2]$, $[\text{Os}(\text{S}_2\text{CH})_2(\text{PPh}_3)_2]$, $[\text{Ir}(\text{S}_2\text{CH})\text{Cl}_2(\text{PPh}_3)_2]$ and $[\text{Ir}(\text{S}_2\text{CH})\text{H}_2(\text{PPh}_3)_2]$, respectively. Each of these products shows IR [$\nu(\text{HCS}_2)$ ca. 1215-1235 w and 900-930 s cm^{-1}] and proton NMR spectra [HCS_2 multiplet τ ca. 0 to -5] characteristic of chelate dithioformate ligands (see Table 1). In contrast the complexes $[\text{RuH}_2(\text{CO})(\text{PPh}_3)_3]$, $[\text{RuHCl}(\text{PPh}_3)_3]$ and $[\text{RhH}_2\text{Cl}(\text{PPh}_3)_3]$ react to afford sulphur-containing products, currently under investigation, which do not display the spectral characteristics of dithioformate ligands but show instead very broad, strong IR bands at ca. 1280-1320 and/or 1000 cm^{-1} . Finally, under the conditions

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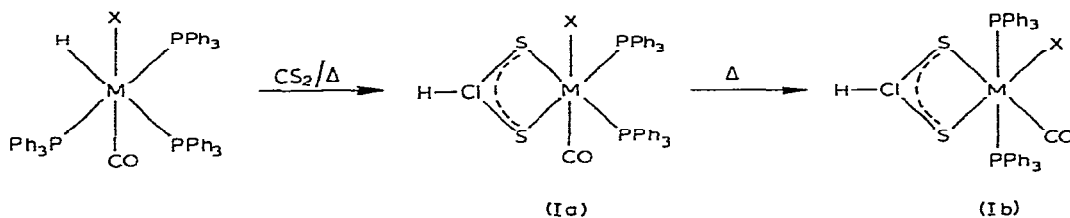
TABLE 1
 INFRARED AND NMR DATA FOR DITHIOFORMATO COMPLEXES

Complex	$\nu(\text{S}_2\text{CH})$ (cm^{-1})	$\tau(\text{S}_2\text{CH})$ (ppm)	$^4J(\text{PH})_{\text{trans}}$ (Hz)	$^4J(\text{PH})_{\text{cis}}$ (Hz)
$\text{Ru}(\text{S}_2\text{CH})\text{Cl}(\text{CO})(\text{PPh}_3)_2$ (Ia)	1230, 925, 920	-1.87	8.0	—
$\text{Ru}(\text{S}_2\text{CH})\text{Cl}(\text{CO})(\text{PPh}_3)_2$ (Ib)	1225, 1220, 930 (935)	-0.25	—	3.0
$\text{Ru}(\text{S}_2\text{CH})\text{Br}(\text{CO})(\text{PPh}_3)_2$ (Ia)	1235 925	-1.70	8.0	—
$\text{Ru}(\text{S}_2\text{CH})\text{Br}(\text{CO})(\text{PPh}_3)_2$ (Ib)	1225 925	-0.15	—	3.0
$\text{Os}(\text{S}_2\text{CH})\text{Cl}(\text{CO})(\text{PPh}_3)_2$ (Ia)	1235 920	-3.65	8.0	—
$\text{Os}(\text{S}_2\text{CH})\text{Cl}(\text{CO})(\text{PPh}_3)_2$ (Ib)	1235 920	-1.85	—	3.0
$\text{Os}(\text{S}_2\text{CH})\text{Br}(\text{CO})(\text{PPh}_3)_2$ (Ia)	1230 920	-3.45	8.5	—
$\text{Os}(\text{S}_2\text{CH})\text{Br}(\text{CO})(\text{PPh}_3)_2$ (Ib)	1222 920	-1.90	—	3.0
$\text{Ru}(\text{S}_2\text{CH})(\text{OCOCF}_3)(\text{CO})(\text{PPh}_3)_2$ (Ib)	1228 930	+0.05	—	2.75
$\text{Os}(\text{S}_2\text{CH})(\text{OCOCF}_3)(\text{CO})(\text{PPh}_3)_2$ (Ia)	1230 925	-3.58	8.25	—
$\text{Ru}(\text{S}_2\text{CH})_2(\text{PPh}_3)_2$ (II) ^a	1220 920, 915, 908	-1.5	1.5	~0
$\text{Os}(\text{S}_2\text{CH})_2(\text{PPh}_3)_2$ (II) ^a	1220, 1215 910, 902	-3.24	5.5	~0
$\text{Ir}(\text{S}_2\text{CH})\text{Cl}_2(\text{PPh}_3)_2$ (III)	1230 915	-3.20	10.0	—
$\text{Ir}(\text{S}_2\text{CH})\text{H}_2(\text{PPh}_3)_2$ (IV) ^b	1235 930	-3.37	—	ca. 3

^a These spectra show additional IR bands at ca. 775 cm^{-1} . ^b IrH, τ 29.93 ppm; $^2J(\text{PH})_{\text{cis}}$ 17.0 Hz; $^4J(\text{HH})_{\text{trans}}$ 1.5 Hz.

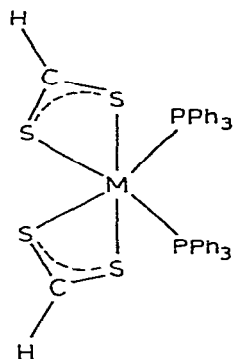
employed in these reactions, the complexes $[\text{OsH}_2(\text{CO})(\text{PPh}_3)_3]$ and *fac*- $[\text{IrH}_3(\text{EPh}_3)_3]$ (E = As, P) appear inert to carbon disulphide.

The stereochemistry of the new dithioformato complexes can be unambiguously established by their proton NMR spectra. Thus the products $[\text{M}(\text{S}_2\text{CH})\text{X}(\text{CO})(\text{PPh}_3)_2]$, formed under mild conditions, show low-field proton NMR triplets [$^4J(\text{PH})_{\text{trans}}$ ca. 8 Hz] indicative of stereochemistry type Ia. The isomeric species $[\text{M}(\text{S}_2\text{CH})\text{X}(\text{CO})(\text{PPh}_3)_2]$, obtained using more vigorous conditions and/or longer reaction times, display low-field triplets [$^4J(\text{PH})_{\text{cis}}$ ca. 3 Hz] consistent with stereochemistry Ib. These assignments are in accord with the expected reaction sequence:

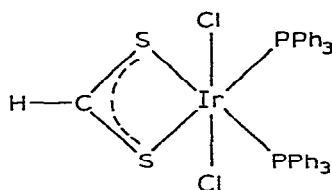


Likewise the osmium complex $[\text{Os}(\text{S}_2\text{CH})(\text{OCOCF}_3)(\text{CO})(\text{PPh}_3)_2]$ [$^4J(\text{PH})_{\text{trans}}$ 8.25 Hz] has stereochemistry Ia (X = OCOCF₃) whereas the more labile ruthenium analogue [$^4J(\text{PH})_{\text{cis}}$ 2.75 Hz] possesses stereochemistry Ib (X = OCOCF₃). The bis(dithioformato) complexes $[\text{M}(\text{S}_2\text{CH})_2(\text{PPh}_3)_2]$ have relatively small P—H couplings [$^4J(\text{PH})_{\text{trans}}$ ca. 1.5–5.5, $^4J(\text{PH})_{\text{cis}} \approx 0$ Hz] and each therefore displays only a narrow doublet at low field. However, this is sufficient to confirm the *cis* stereochemistry II previously established for the ruthenium derivative by X-ray diffraction methods [4]. Finally, the low-field triplets [$^4J(\text{PH})_{\text{trans}}$ 10.0 Hz] is indicative of stereochemistry III for $[\text{Ir}(\text{S}_2\text{CH})\text{Cl}_2(\text{PPh}_3)_2]$ whereas the triplet of triplets pattern [τ -3.37, $^4J(\text{PH})_{\text{cis}}$ ca. 3, $^4J(\text{HH})_{\text{trans}}$ 1.5 Hz] observed for the dihydride complex $[\text{Ir}(\text{S}_2\text{CH})\text{H}_2(\text{PPh}_3)_2]$ establishes stereochemistry IV and is

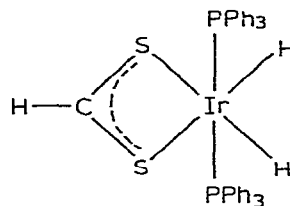
fully consistent with the high-field proton pattern [τ 29.93 triplet of doublets, $^2J(\text{PH})_{\text{cis}}$ 17.0, $^4J(\text{HH})_{\text{trans}}$ 1.5 Hz].



(II)



(III)



(IV)

These data amply demonstrate that NMR couplings of the dithioformate proton are a valuable source of stereochemical information. Furthermore, equivalent coupling of the dithioformate proton to two ^{31}P nuclei (Ia and Ib) or two ^1H nuclei (IV) clearly favours the symmetrical structure shown for the dithioformate ligands in I-IV and virtually eliminates the asymmetric alternatives:



References

- 1 I.S. Butler and A.E. Fenster, *J. Organometal. Chem.*, 66 (1974) 161 and ref. therein.
- 2 R.O. Harris, N.K. Hota, L. Sadavoy and J.M.C. Yuen, *J. Organometal. Chem.*, 54 (1973) 259.
- 3 F.G. Moers, R.W.M. Ten Hoedt and J.P. Langhout, *J. Organometal. Chem.*, 65 (1974) 93.
- 4 A.E. Kahnin, A.I. Gusev and Y.T. Struchkov, *J. Structural Chem.*, 14 (1973) 804.

Erratum

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page C14, line 11 should read:

atom. Thus the line width at half height for the C_{α} signal of $[(\text{OC})_9\text{Co}_3\text{CCHC}_9\text{H}_{19}]^+$